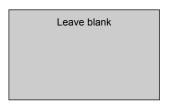
Surname	(Othe	r Names				
Centre Number				Candida	ate Number		
Candidate Signature							



General Certificate of Education January 2006 Advanced Subsidiary Examination

PHYSICS (SPECIFICATION A) PA01 Unit 1 Particles, Radiation and Quantum Phenomena



Thursday 12 January 2006 9.00 am to 10.00 am

For this paper you must have:

- a calculator
- a pencil and a ruler

Time allowed: 1 hour

Instructions

- Use blue or black ink or ball-point pen.
- Fill in the boxes at the top of this page.
- Answer all questions.
- Answer the questions in the spaces provided.
- Show all your working.
- Do all rough work in this book. Cross through any work you do not want marked.

Information

- The maximum mark for this paper is 50. This includes up to 2 marks for the Quality of Written Communication.
- The marks for questions are shown in brackets.
- A *Data Sheet* is provided on pages 3 and 4. You may wish to detach this perforated sheet at the start of the examination.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers. Questions indicated on the paper should be answered in continuous prose. Quality of Written Communication will be assessed in these answers.

For Examiner's Use							
Number	Mark	Number	Mark				
1							
2							
3							
4							
5							
6							
Total (Co	lumn 1)	-					
Total (Co	lumn 2) —	-					
Quality of Written Communication							
TOTAL							
Examiner	's Initials						

Data Sheet

- A perforated *Data Sheet* is provided as pages 3 and 4 of this question paper.
- This sheet may be useful for answering some of the questions in the examination.
- You may wish to detach this sheet before you begin work.

	Fundamental constants and values							
	Quantity	Symbol	Value	Units				
	speed of light in vacuo	c	3.00×10^{8}	m s ⁻¹				
	permeability of free space	μ_0	$4\pi \times 10^{-7}$	H m ⁻¹				
	permittivity of free space	ϵ_0	8.85×10^{-12}	F m ⁻¹				
	charge of electron	e	1.60×10^{-19}	C				
	the Planck constant	h	6.63×10^{-34}	J s				
	gravitational constant	G	6.67×10^{-11}	$N m^2 kg^{-2}$				
	the Avogadro constant	$N_{\rm A}$	6.02×10^{23}	mol^{-1}				
	molar gas constant	R	8.31	J K ⁻¹ mol				
	the Boltzmann constant	k	1.38×10^{-23}	J K ⁻¹				
	the Stefan constant	σ	5.67×10^{-8}	$W m^{-2} K^{-4}$				
	the Wien constant	α	2.90×10^{-3}	m K				
	electron rest mass	$m_{\rm e}$	9.11×10^{-31}	kg				
	(equivalent to 5.5×10^{-4} u)							
	electron charge/mass ratio	e/m _e	1.76×10^{11}	C kg ⁻¹				
	proton rest mass	$m_{ m p}$	1.67×10^{-27}	kg				
Ī	(equivalent to 1.00728u)		_					
	proton charge/mass ratio	$e/m_{\rm p}$	9.58×10^{7}	C kg ⁻¹				
ļ	neutron rest mass	$m_{\rm n}$	1.67×10^{-27}	kg				
	(equivalent to 1.00867u)							
İ	gravitational field strength	g	9.81	N kg ⁻¹				
ļ	acceleration due to gravity	g	9.81	m s ⁻²				
	atomic mass unit	u	1.661×10^{-27}	kg				
I	(1u is equivalent to							
	931.3 MeV)	1						

Fundamental particles

	-		
Class	Name	Symbol	Rest energ
			/MeV
photon	photon	γ	0
lepton	neutrino	$ u_{ m e}$	0
		$ u_{\mu}$	0
	electron	e [±]	0.510999
	muon	μ^{\pm}	105.659
mesons	pion	π^{\pm}	139.576
		π^0	134.972
	kaon	\mathbf{K}^{\pm}	493.821
		K^0	497.762
baryons	proton	p	938.257
	neutron	n	939.551

Properties of quarks

Type	Charge	Baryon number	Strangeness
u	$+\frac{2}{3}$	$+\frac{1}{3}$	0
d	$-\frac{1}{3}$	$+\frac{1}{3}$	0
S	$-\frac{1}{3}$	$+\frac{1}{3}$	-1

Geometrical equations

 $arc\ length = r\theta$ $circumference\ of\ circle=2\pi r$ area of circle = πr^2 area of cylinder = $2\pi rh$ *volume of cylinder* = $\pi r^2 h$ area of sphere = $4\pi r^2$ *volume of sphere* = $\frac{4}{3}\pi r^3$

Mechanics and Applied

Physics
$$v = u + at$$

$$s = \left(\frac{u + v}{2}\right)t$$

$$s = ut + \frac{at^2}{2}$$

$$v^2 = u^2 + 2as$$

$$F = \frac{\Delta(mv)}{\Delta t}$$

$$P = Fv$$
efficiency = \frac{power output}{power input}
$$\omega = \frac{v}{r} = 2\pi f$$

$$a = \frac{v^2}{r} = r\omega^2$$

$$I = \sum mr^2$$

$$E_{\mathbf{k}} = \frac{1}{2} I \omega^2$$

$$\omega_2 = \omega_1 + \alpha t$$

$$\theta = \omega_1 t + \frac{1}{2} \alpha t^2$$

$$\omega_2^2 = \omega_1^2 + 2\alpha\theta$$

$$\theta = \frac{1}{2} \left(\omega_1 + \omega_2 \right) t$$

$$T = I\alpha$$

angular momentum = $I\omega$ $W = T\theta$ $P = T\omega$

angular impulse = change of $angular\ momentum = Tt$ $\Delta Q = \Delta U + \Delta W$ $\Delta W = p\Delta V$ pV^{γ} = constant

work done per cycle = area of loop

input power = calorific value × fuel flow rate

indicated power as (area of $p - V \mid P = I^2 R$ $loop) \times (no.\ of\ cycles/s) \times$ (no. of cylinders)

friction power = indicated power – brake power

efficiency =
$$\frac{W}{Q_{\rm in}} = \frac{Q_{\rm in} - Q_{\rm out}}{Q_{\rm in}}$$

maximum possible

$$efficiency = \frac{T_{\rm H} - T_{\rm C}}{T_{\rm H}}$$

Fields, Waves, Quantum Phenomena

$$g = \frac{F}{m}$$

$$g = -\frac{GM}{r^2}$$

$$g = -\frac{\Delta V}{\Delta x}$$

$$V = -\frac{GM}{r}$$

$$a = -(2\pi f)^2 x$$

$$v = \pm 2\pi f \sqrt{A^2 - x^2}$$

$$x = A \cos 2\pi f t$$

$$T = 2\pi \sqrt{\frac{I}{g}}$$

$$\lambda = \frac{\omega s}{D}$$

$$d \sin \theta = n\lambda$$

$$\theta \approx \frac{\lambda}{D}$$

$$1^{n_2} = \frac{\sin \theta_1}{\sin \theta_2} = \frac{c_1}{c_2}$$

$$1^{n_2} = \frac{n_2}{n_1}$$

$$\sin \theta_c = \frac{1}{n}$$

$$E = hf$$

$$hf = \phi + E_k$$

$$hf = E_1 - E_2$$

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

$$c = \frac{1}{\sqrt{u_0 \varepsilon_0}}$$

Electricity

$$\begin{aligned}
&\in = \frac{E}{Q} \\
&\in = I(R+r) \\
&\frac{1}{R_{T}} = \frac{1}{R_{1}} + \frac{1}{R_{2}} + \frac{1}{R_{3}} + \cdots \\
&R_{T} = R_{1} + R_{2} + R_{3} + \cdots \\
&P = I^{2}R \\
&E = \frac{F}{Q} = \frac{V}{d} \\
&E = \frac{1}{4\pi\epsilon_{0}} \frac{Q}{r^{2}} \\
&E = \frac{1}{2} QV \\
&F = BII \\
&F = BQv
\end{aligned}$$

 $Q = Q_0 e^{-t/RC}$

 $\Phi = BA$

Turn over

magnitude of induced e.m.f. = $N \frac{\Delta \Phi}{\Delta t}$

$$I_{\rm rms} = \frac{I_0}{\sqrt{2}}$$

$$V_{\rm rms} = \frac{V_0}{\sqrt{2}}$$

Mechanical and Thermal Properties

the Young modulus =
$$\frac{\text{tensile stress}}{\text{tensile strain}} = \frac{F}{A} \frac{l}{e}$$

energy stored =
$$\frac{1}{2}$$
 Fe

$$\Delta Q = mc \ \Delta \theta$$

$$\Delta Q = ml$$

$$pV = \frac{1}{3} Nm\overline{c^2}$$

$$\frac{1}{2}m\overline{c^2} = \frac{3}{2}kT = \frac{3RT}{2N_A}$$

Nuclear Physics and Turning Points in Physics

$$force = \frac{eV_{p}}{d}$$

$$force = Bev$$

radius of curvature =
$$\frac{mv}{Be}$$

$$\frac{eV}{d} = mg$$

 $work\ done = eV$

$$F = 6\pi \eta r v$$

$$I = k \frac{I_0}{r^2}$$

$$\frac{\Delta N}{\Delta t} = -\lambda N$$

$$\lambda = \frac{h}{\sqrt{2meV}}$$

$$N = N_0 e^{-\lambda t}$$

$$T_{\frac{1}{2}} = \frac{\ln 2}{\lambda}$$

$$R = r_0 A^{\frac{1}{3}}$$

$$E = mc^2 = \frac{m_0 c^2}{\left(1 - \frac{v^2}{c^2}\right)^{\frac{1}{2}}}$$

$$l = l_0 \left(1 - \frac{v^2}{c^2} \right)^{\frac{1}{2}}$$

$$t = \frac{t_0}{\left(1 - \frac{v^2}{c^2}\right)^{\frac{1}{2}}}$$

Astrophysics and Medical Physics

Body Mass/kg Mean radius/m

Sun
$$2.00 \times 10^{30}$$
 7.00×10^{8}
Earth 6.00×10^{24} 6.40×10^{6}

1 astronomical unit = 1.50×10^{11} m

1 parsec =
$$206265 \text{ AU} = 3.08 \times 10^{16} \text{ m} = 3.26 \text{ ly}$$

1 light year = 9.45×10^{15} m

Hubble constant $(H) = 65 \text{ km s}^{-1} \text{ Mpc}^{-1}$

angle subtended by image at eye $M = \frac{M}{M}$

angle subtended by object at unaided eye

$$M = \frac{f_{\rm o}}{f_{\rm c}}$$

$$m - M = 5 \log \frac{d}{10}$$

 $\lambda_{\text{max}}T = \text{constant} = 0.0029 \text{ m K}$

v = Hd

$$P = \sigma A T^4$$

$$\frac{\Delta f}{f} = \frac{v}{c}$$

$$\frac{\Delta\lambda}{\lambda} = -\frac{\nu}{c}$$

$$R_{\rm s} \approx \frac{2GM}{c^2}$$

Medical Physics

 $power = \frac{1}{f}$

$$\frac{1}{u} + \frac{1}{v} = \frac{1}{f}$$
 and $m = \frac{v}{u}$

intensity level = $10 \log \frac{I}{I_0}$

 $I = I_0 e^{-\mu t}$

$$\mu_{\rm m} = \frac{\mu}{\alpha}$$

Electronics

Resistors

Preferred values for resistors (E24) Series: 1.0 1.1 1.2 1.3 1.5 1.6 1.8 2.0 2.2 2.4 2.7 3.0 3.3 3.6 3.9 4.3 4.7 5.1 5.6 6.2 6.8 7.5 8.2 9.1 ohms and multiples that are ten times greater

$$Z = \frac{V_{\rm rms}}{I_{\rm rms}}$$

$$\frac{1}{C_{\rm T}} = \frac{1}{C_1} + \frac{1}{C_2} + \frac{1}{C_3} + \cdots$$

$$C_{\mathrm{T}} = C_1 + C_2 + C_3 + \cdots$$

$$X_{\rm C} = \frac{1}{2\pi fC}$$

Alternating Currents

$$f = \frac{1}{T}$$

Operational amplifier

$$G = \frac{V_{\text{out}}}{V_{\text{in}}}$$
 voltage gain

$$G = -\frac{R_{\rm f}}{R_{\rm 1}}$$
 inverting

$$G = 1 + \frac{R_{\rm f}}{R_1}$$
 non-inverting

$$V_{\text{out}} = -R_{\text{f}} \left(\frac{V_1}{R_1} + \frac{V_2}{R_2} + \frac{V_3}{R_3} \right)$$
 summing

Answer all questions in the spaces provided.

A rac	dioactive isotope of carbon is represented by ${}^{14}_{6}$ C.
(a)	Using the same notation, give the isotope of carbon that has two fewer neutrons.
(b)	Calculate the charge on the ion formed when \mathbf{two} electrons are removed from an atom of $^{14}_{\ 6}$ C.
(c)	Calculate the value of $\frac{\text{charge}}{\text{mass}}$ for the nucleus of an atom of $^{14}_{6}$ C.
	(2 marks)

Turn over for the next question

1

2	(a)	One quantity in the photoelectric equation is a characteristic property of the metal that emits photoelectrons. Name and define this quantity.
		(2 marks)
	(b)	A metal is illuminated with monochromatic light. Explain why the kinetic energy of the photoelectrons emitted has a range of values up to a certain maximum.
		You may be awarded marks for the quality of written communication in your answer.
		(3 marks)

(c)	A gold surface is illuminated with monochromatic ultraviolet light of frequency $1.8 \times 10^{15}\mathrm{Hz}$. The maximum kinetic energy of the emitted photoelectrons is $4.2 \times 10^{-19}\mathrm{J}$. Calculate, for gold,					
	(i)	the work function, in J,				
	(ii)	the threshold frequency.				
		(5 marks)				

Turn over for the next question

Turn over ▶

3	(a)	(i)	Give an example of an exchange particle other than a W^+ or W^- particle, and state the fundamental force involved when it is produced. exchange particle
		(ii)	State what roles exchange particles can play in an interaction.
			(4 marks)
	(b)	Fron	n the following list of particles,
		p	\overline{n} ν_e e^+ $\mu^ \pi^0$
		ident	rify all the examples of
		(i)	hadrons,
		(ii)	leptons,
		(iii)	antiparticles,
		(iv)	charged particles. (4 marks)

4	(i)	A negative muon, μ^- , is 207 times more massive than an electron. Calculate the de Broglie wavelength of a negative muon travelling at $3.0 \times 10^6\text{m}\text{s}^{\text{-1}}$.
	(ii)	Using values from the data sheet calculate the ratio $\frac{\text{rest mass of } \pi^0}{\text{rest mass of } \mu^-}$ where π^0 is a neutral pion.
	(iii)	Calculate the speed necessary for a π^0 to have the same de Broglie wavelength as that of the μ^- in part (i).
		(6 marks)

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5 Figure 1 shows a ray of light passing from air into glass at the top face of glass block 1 and emerging along the bottom face of glass block 2.

refractive index of the glass in block 1 = 1.45

Figure 1

air θ_1 incident ray $\theta_2 = 15.5^{\circ}$ glass block 2

glass block 1

air

emergent ray

(a)	Calculate							
	(i)	the incident angle θ_1 ,						
	(ii)	the refractive index of the glass in block 2,						
	····							
	(iii)	the angle $ heta_3$ by considering the refraction at point ${f A}$.						
		(7 marks)						
(b)	In which of the two blocks of glass will the speed of light be greater?							
	Expl	ain your reasoning.						
		(2 marks)						
(c)		g a ruler, draw the path of a ray partially reflected at A on Figure 1 . Continue the show it emerging into the air. No calculations are expected. (2 marks)						
		(2 marks)						

6			ne energy levels of an atom are shown belonpact.	ow. The atom may be <i>ionised</i> by
			energy/ $10^{-17} \mathrm{J}$	
			0.00	ionisation level
			-1.97	level E
			-2.20	level D level C level B
			-2.43	ievei b
			-4.11	level A (ground state)
	(a)	(i)	State what is meant by the ionisation of	an atom.
		(ii)	Calculate the minimum kinetic energy, in ionise the atom from its ground state.	n eV, of an incident electron that could
				(2 marks)

part	may be awarded marks for the quality of written communication in your answer to s (b)(i) and (b)(ii).	
The	atom in the ground state is given 5.00×10^{-17} J of energy by electron impact.	
(i)	State what happens to this energy.	
(ii)	Describe and explain what could happen subsequently to the electrons in the higher energy levels.	
	(4 marks)	
	(4 marks) ntify two transitions between energy levels that would give off electromagnetic ation of the same frequency.	
radi	ntify two transitions between energy levels that would give off electromagnetic	
radi	ntify two transitions between energy levels that would give off electromagnetic ation of the same frequency.	
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radi ——	ntify two transitions between energy levels that would give off electromagnetic ation of the same frequency.	
radi ——	to to (2 marks)	

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